



WORKSHOP: VALORISATION OF TRADITIONAL PROCESSING OF INDIGENOUS AND UNDERUTILISED FRUITS

Institute of Technology of Cambodia, Phnomhnh, Cambodia

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PURIFICATION OF ANTHOCYANIN BY POLYMERIC ADSORBENT RESINS

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Introduction

- There is a general tendency nowadays toward using **natural coloring** in foods as a substitute to chemical food coloring that has harmful physiological effects.
- Studies are being done accordingly to develop food coloring products from natural sources such as barberry, blackberry...the pigment of which contain various **phenolic compounds/anthocyanins**.
- Prices for anthocyanins or other polyphenol extracts cover a wide range depending on the **concentration of active components**,costs into the hundreds of dollars/ kilogram.

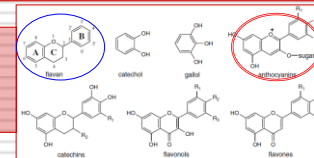
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Major families of phenolic compounds

The typical structural characteristic shared by most polyphenols: the three-membered flavan ring system.

Their role in fruits/ vegetables:

- responsible for structural and protective functions.
- contributing to flavour, color, astringency and bitterness.



Their role in human health: based on their antioxidant activity against reactive species involved in aging and in chronic, autoimmune, inflammatory, coronary, cancer and degenerative diseases.

Nathan, 2009

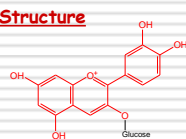
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Anthocyanin molecules

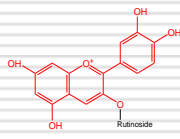
Characteristics

- Belong to the flavonoid group of polyphenol
- Water soluble pigments, responsible for the **red, purple, and blue** colors of fruits, vegetables, and flowers.

Structure



Cyanidin-3-glucoside



Cyanidin-3-rutinoside

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Antioxidant interest: problematic research

ORAC value apply to water-soluble antioxidants

The higher the number, the stronger the antioxidant properties.

	Açaí	Blackberry	Strawberry	Raspberry	Red dragon	White dragon
ORAC value	48,6 - 61,5	13,7 - 25,1	18,3 - 22,9	19,2 - 22,6	7,59-10,76	2,96-5,23

ORAC = (Oxygen Radical Absorbance Capacity), (µmol eq trolox /g)
(Del Pozo Insfran, D., 2007, USDA, 2004)

The USDA recommends the consuming of 3,000 to 5,000 ORAC units/day.

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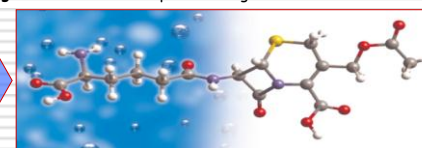
Separation process

Conventional process

High T°: leads to color loss, Maillard reactions since extracts may contain high amounts of sugar.

Organic solvent: causes pseudoallergic reactions in humans.

Absorption-desorption process



- minimizes degradation, simplicity of design, operation and scale up,
- ease of regeneration and
- low cost.

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Major types of adsorbents

- Activated carbon;
- Mineral adsorbents include siliceous materials, clay and natural zeolites; and
- **polymeric adsorbents (resin).**

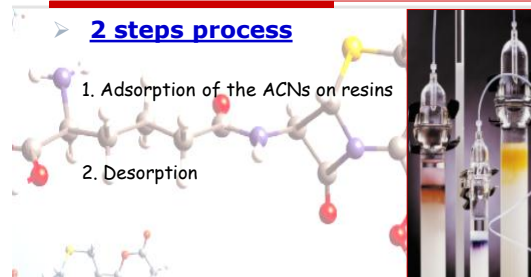
RESIN

- Approved by for food contact use, by the European code of regulation.
- Most of the impurities are removed.

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Resin purification process

2 steps process



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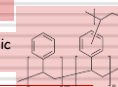
Step 1: Which type of Resin could be used for anthocyanin purification?

- **Nature of the solvent:** polar, non-polar
- **Functionality of solute:** aromatic, high electron density, saturated C-C bond
- **Polarisation:** the solute has groups that can be easily polarised or not.
 - Easily polarised: $-CO_2H$, $-NH_2$
 - Can be polarised under extreme conditions: $>C=O$ or ethers,
- **Size of solute:**
 - Small molecule: molecular < 1000 D,
 - large molecule: molecular > 10,000 D

Available Resins:

- Polystyrene-divinylbenzene copolymers,
- Polymethacrylate,
- Divinylbenzene-ethylvinylbenzene copolymers, and
- Vinylpyridine.

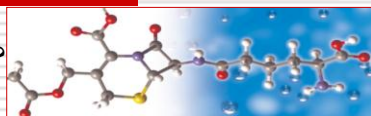
It's a **Polystyrene-divinylbenzene copolymers**, with a specific surface area, porosity, and pore radius ($C_{24}H_{24}$).



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Step 2: Which solvent for desorption of anthocyanins?

- Alcohols:
 - Methanol ?
 - Ethanol ?
 - Propanol ?

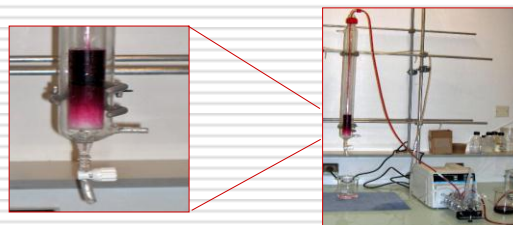


- **Methanol** : more efficient (96% to 100% of anthocyanins could be recovered) in desorbing anthocyanins from the polystyrene-divinylbenzene copolymerisate than ethanol and 2-propanol.

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Adsorption of anthocyanins on resins- Step 1.

Hydrophobic interaction, hydrogen bonding



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Desorption of anthocyanins from resins- Step 2.

With an organic solvent which can weaken the attractive forces between the solute and the polymer for removing ACNs



At the beginning



At the end

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Result of ACN's purification with resin

		Juice	Final Extract	Yield ACN % (extract/juice)	Yield DM % (extract/juice)	Purity of final extract (ACN/DM)
BBerry	ACN contents (mg/100mL)	40,8	130,8	96	6,05	19,46%
	E ^{1%} _{1cm}	4	36			
	VI/BI	24,6	34,1			
Açai	ACN contents (mg/100mL)	31,2	54,3	87	31,32	8%
	E ^{1%} _{1cm}	17,8	48			
	VI/BI	54,4	51			

DM = Dry matter ACN = Anthocyanins VI = Violet Index; BI = Brown Index
E^{1%}_{1cm} = Strength colour : correspond to the Abs of (x g) pigment in 100mL buffer pH3

- **Simplicity** , - **Great % of ACN's recovery**
- **Increase of the colour strength**, - **No colour degradation**

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Research Proposal

Extraction and purification of Polyphenols from cashew apples waste and of betalains from pitaya by-products



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Cashew apple phenolics

- **1978**: Satyanarayana, extracted, purified and characterised three flavonols from an **Indian** variety, i.e. quercetin 3-O-galactoside, myricetin, and quercetin.
- **2001**, Moura & **2007** de Abreu, colorimetrically measured yellow flavonoids and anthocyanins from **Brazilian** variety... without further characterisation.
- **2009**, Laetitia extracted Monomeric phenols by acetone/water (60:40) from the **skin and flesh** of four cashew apple from **Brazil and Bénin** (West Africa).
- **2012**, ADOU, phenolic profile of the two varieties of cashew (anacardium occidentale L.) cultivated in **Côte d'Ivoire**.

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Cashew apple phenolics

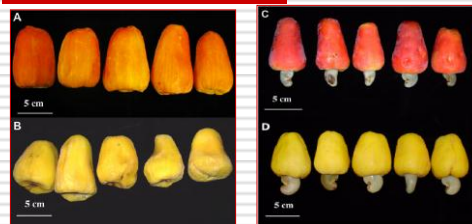


Figure 1. Cashew apple from Brazil (A-B), cashew apple from Benin (C-D). (Laetitia, 2009).

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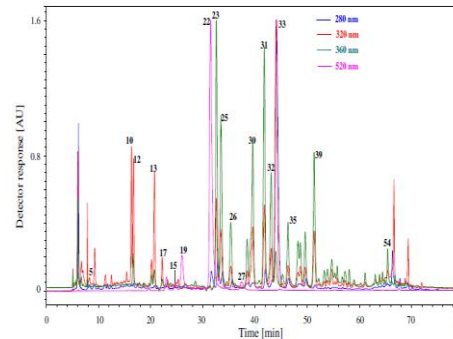


Fig. 2. HPLC chromatograms of skin extract of cashew apple (Parake Rouge variety): (5) gallic acid; (12, 13) p-coumaric acid conjugates; (15) cyanidin hexoside; (19) peridin hexoside; (22) peonidin hexoside; (23) myricetin hexoside; (25) myricetin hexoside; (26) myricetin pentoside; (27) unknown anthoxyanidin hexoside; (30) myricetin 3-O-rhamnoside; (31) quercetin 3-O-galactoside; (32) quercetin 3-O-glucoside; (35) quercetin pentoside; (39) quercetin 3-O-rhamnoside; (54) quercetin. Some unknown peaks are also numbered (10, 17, and 33). Peak heights were normalised at 1.6 AU at $\lambda = 280$ nm.

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Cashew apple phenolics

- Skins from red Benin variety and organe Brazil: **anthocyanidin glycosides**.
- Skins from the apples were **1520 times richer in total phenolic compounds than their fleshs**, reaching **30→110 and 2→5 mg/100 g fresh weight**, respectively.
- African apples: be richer in skin flavonols and flavonol glycosides than Brazilian ones, while the reverse situation was found in the flesh.

Dragon fruit phenolics

- *Hylocereus undatus* :white dragon fruit and *Hylocereus polyrhizus* : red dragon fruit.
- Total phenolic content (TPC) assay demonstrated that **peels** of both *Hylocereus* species **contained higher phenolic** content than the **pulps**.
 - The phenolic content in **peels** of *white dragon* was **higher than red dragon**,
 - but the phenolic content in **pulps** of *white dragon* was **much lower than red dragon**.

Nurliyana, 2010

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Dragon fruit phenolics

in order to gain better views on the antioxidant level and activities in *Hylocereus* species,...

→ further studies on **purification, identification and quantification** of **each phenolic compound and other non-phenolic compounds such as carotenoids and betalains**

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Extraction technology and Purification Process

- Extraction techniques, which have been replacing conventional ones, include:
 - supercritical fluid extraction (SFE),
 - pressurized liquid extraction (PLE),
 - microwave-assisted extraction (MAE) and
 - ultrasound-assisted extraction (UAE).
 - Membrane extraction

These alternative techniques reduce considerably the use of solvents and accelerate the extraction process.

- Purification polymeric adsorber resin: polystyrene-divinylbenzene copolymer (PSDVB).

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Thank you for your welcome and for your listening

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